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 Answer: Manganese belongs to group number 7 and its oxidation state in KMnO_4 is +7 i.e., KMnO_4 $1 + x + 4(-2) = 0$
 $\Rightarrow 1 + x - 8 = 0 \Rightarrow x = 7$
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 oxo-anion of Manganese
 (Mn) in which it shows the
 oxidation state equal to
 its group number. [1]
 Answer: Manganese
 belongs to group number
 7 and its oxidation state
 in KMnO_4 is +7 i.e.,
 KMnO_4 $1 + x + 4(-2) = 0$
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